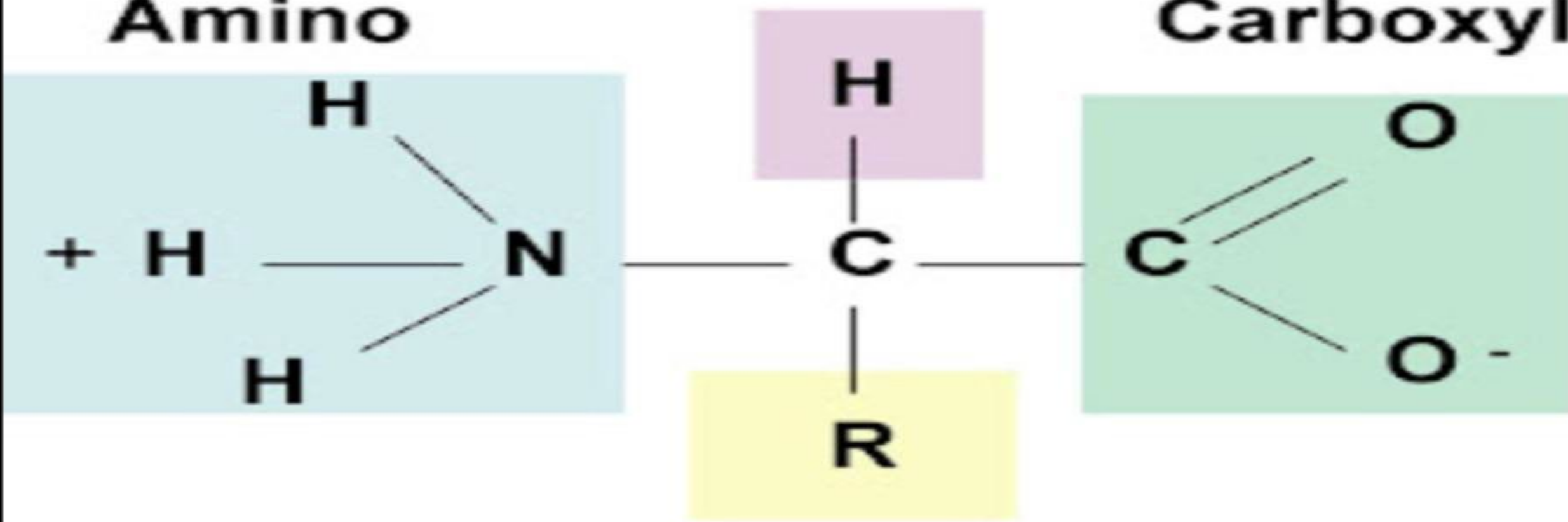


Amino Acid Structure

Hydrogen

Amino

Carboxyl



R-group

Amino Acid	Symbol	Structure
Cl. 1 : Amino Acids with Aliphatic R-Groups		
Glycine	Gly - G	$\begin{array}{c} \text{H} - \text{CH} - \text{COOH} \\ \\ \text{NH}_2 \end{array}$
Alanine	Ala - A	$\begin{array}{c} \text{CH}_3 - \text{CH} - \text{COOH} \\ \\ \text{NH}_2 \end{array}$
Valine	Val - V	$\begin{array}{c} \text{H}_3\text{C} \diagup \\ \text{H}_3\text{C} \diagdown \end{array} \text{CH} - \underset{\substack{ \\ \text{NH}_2}}{\text{CH}} - \text{COOH}$
Leucine	Leu - L	$\begin{array}{c} \text{H}_3\text{C} \diagup \\ \text{H}_3\text{C} \diagdown \end{array} \text{CH} - \text{CH}_2 - \underset{\substack{ \\ \text{NH}_2}}{\text{CH}} - \text{COOH}$
Isoleucine	Ile - I	$\begin{array}{c} \text{H}_3\text{C} - \text{CH}_2 \diagup \\ \text{H}_3\text{C} \diagdown \end{array} \text{CH} - \underset{\substack{ \\ \text{NH}_2}}{\text{CH}} - \text{COOH}$
Cl. 2 : Non-Aromatic Amino Acids with Hydroxyl R-Groups		
Serine	Ser - S	$\text{HO} - \text{CH}_2 - \underset{\substack{ \\ \text{NH}_2}}{\text{CH}} - \text{COOH}$
Threonine	Thr - T	$\begin{array}{c} \text{H}_3\text{C} \diagup \\ \text{HO} \diagdown \end{array} \text{CH} - \underset{\substack{ \\ \text{NH}_2}}{\text{CH}} - \text{COOH}$

Cl. 3 : Amino Acids with Sulfur-Containing R-Groups

Cysteine	Cys - C	$\text{HS} - \text{CH}_2 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH}$
Methionine	Met-M	$\text{H}_3\text{C} - \text{S} - (\text{CH}_2)_2 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH}$

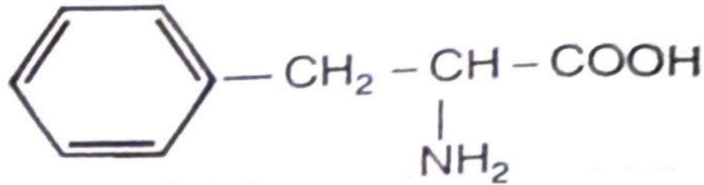
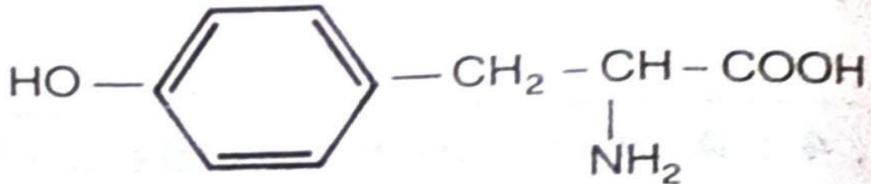
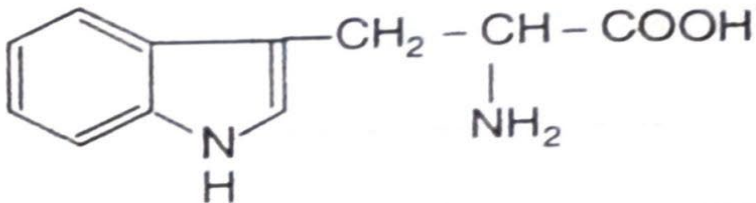
Cl. 4 : Acidic Amino Acids and their Amides

Aspartic Acid	Asp - D	$\text{HOOC} - \text{CH}_2 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH}$
Asparagine	Asn - N	$\text{H}_2\text{N} - \underset{\text{O}}{\underset{ }{\text{C}}} - \text{CH}_2 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH}$
Glutamic Acid	Glu - E	$\text{HOOC} - \text{CH}_2 - \text{CH}_2 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH}$
Glutamine	Gln - Q	$\text{H}_2\text{N} - \underset{\text{O}}{\underset{ }{\text{C}}} - \text{CH}_2 - \text{CH}_2 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH}$

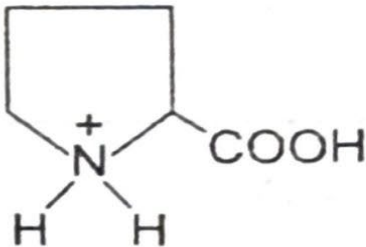
Cl. 5 : Basic Amino Acids

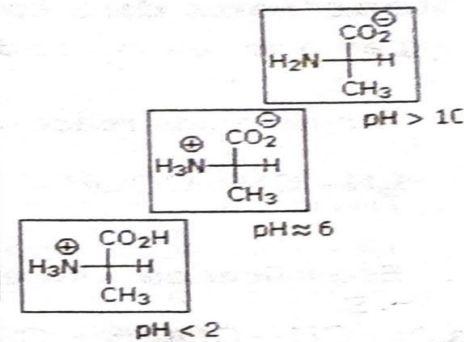
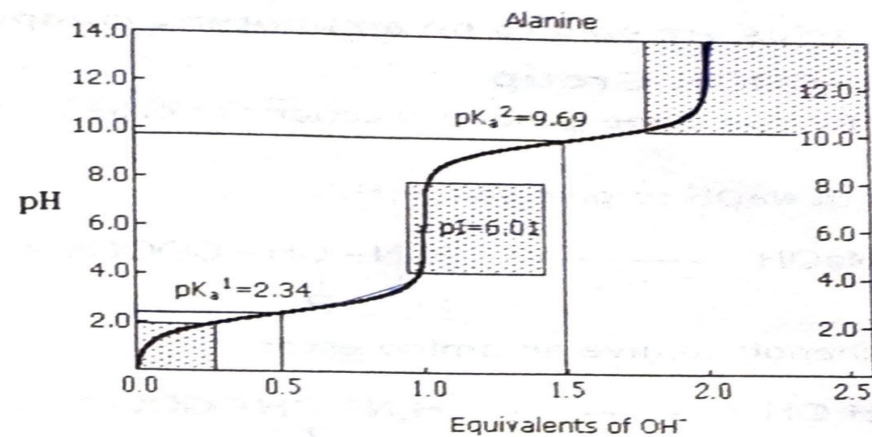
Arginine	Arg - R	$\begin{array}{c} \text{HN} - \text{CH}_2 - \text{CH}_2 - \text{CH}_2 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH} \\ \\ \text{C} = \text{NH} \\ \\ \text{NH}_2 \end{array}$
Lysine	Lys - K	$\text{H}_2\text{N} - (\text{CH}_2)_4 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH}$
Histidine	His - H	$\begin{array}{c} \text{CH}_2 - \underset{\text{NH}_2}{\underset{ }{\text{CH}}} - \text{COOH} \\ \\ \text{Imidazole ring} \end{array}$

Cl. 6 : Amino Acids with Aromatic Rings

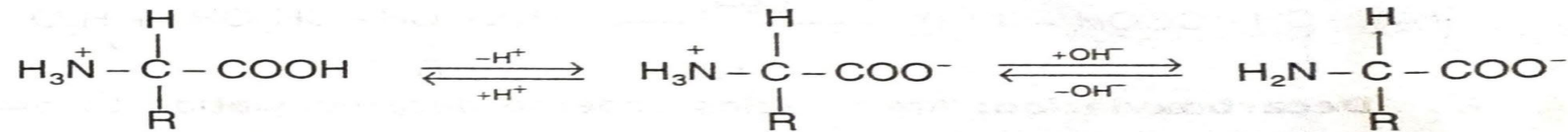
Phenylalanine	Phe - F	
Tyrosine	Tyr - Y	
Tryptophan	Trp - W	

Cl. 7 : Imino Acids

Proline	Pro - P	
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The graph shows that pK of carboxylic group is 2.34 and that of amino group is 9.69. Therefore pI should be 6.01.



Cation of low pH
(acidic pH)

Net charge is +1

Zwitter ion at
isoelectric pH

Net charge is zero

Anion at high pH
(alkaline)

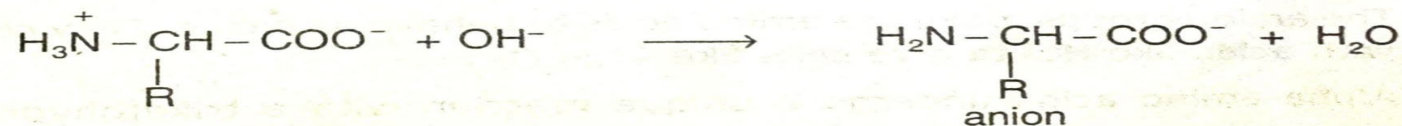
Net charge -1

1. If acid is added to Zwitter ion, a cation is resulted.



Zwitter ion served as a proton acceptor. This indicates the basic property of the amino acid.

2. If a base is added to Zwitter ion, anion is resulted



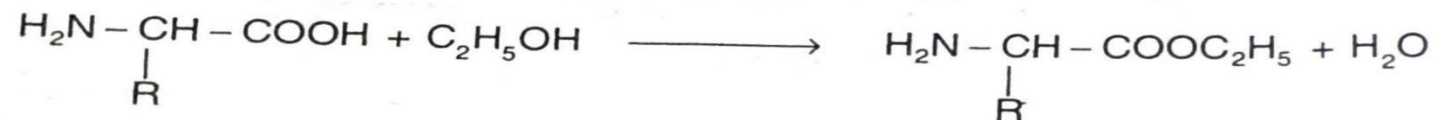
3.3.4 Reactions due to -COOH Group

Being an acid, amino acid follows some general reactions of organic acids. They are:

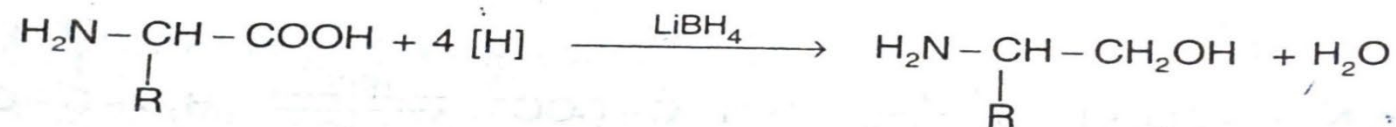
1. Amino acids react with NaOH to give salt + H₂O.



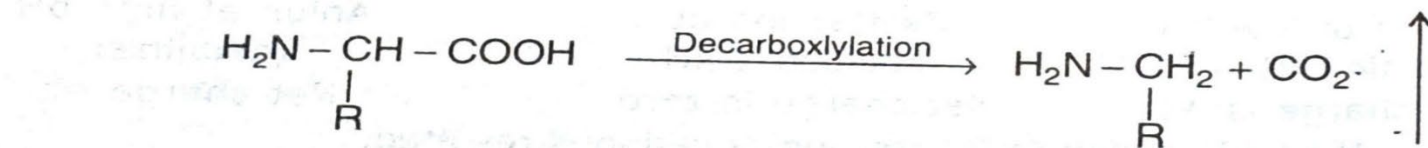
2. Esterification with ethanol; to give an amino ester



3. Amino acids can be reduced to amino alcohol; in presence of pot agent like lithium borohydride or lithium aluminium hydride.



4. **Decarboxylation:** Amino acids undergo decarboxylation to produce corresponding amines.



Some biologically important amines like histamine from histidine; tyramine from tyrosine and γ -amino butyric acid (GABA) from glutamate are produced by decarboxylation process.

5. Some Amino acids + NH₃ \longrightarrow Amides
- Aspartic acid + NH₃ \longrightarrow Asparagine
- Glutamic acid + NH₃ \longrightarrow Glutamine

This reaction is important in the transport of NH₃ in the body.