61513

Fifth Semester B.Sc. Degree Examination, March/April 2021

(CBCS - Semester Scheme)

Biochemistry

Paper V — BIOCHEMISTRY

Time: 3 Hours]

[Max. Marks: 70

- Instructions to Candidates :
 - 1) This paper is for the students of new syllabus: 2014-15.
 - 2) The Question Paper has two Parts: Part-A and Part-B.
 - 3) Answer any EIGHT questions from Part-A.
 - 4) Answer any NINE questions from Part-B.

PART - A

Answer any **EIGHT** of the following questions. Each question carries 2 marks: $(8 \times 2 = 16)$

- 1. What are epimers? Give an example.
- 2. Write the structure of isomaltose.
- 3. What are heteropolysaccharides? Give an example.
- 4. Write any two importance of phospholipids.
- 5. Define saponification number and give its significance.
- 6. What are androgens? Mention any two of their biological functions.
- 7. Write any two biological importance of oxytocin.
- 8. How does an amino acid react with Sanger's reagent?
- 9. What is redox reactions? Give an example.
- 10. Give the relationship between standard free energy change and equilibrium constant.
- 11. What is P:O ratio?
- 12. Write the structure of ATP.



61513

PART - B

Answer any NINE of the following questions	s. Each question carries 6 marks :
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- Explain about plant storage polysaccharide. 13. (a) Write the structure of glucose-6-phosphate. Give its biological importance. (b) (4 + 2)Write the partial structure of any two heteropolysaccharides. 14. (a) (4 + 2)Differentiate between glucaric and glucuronic acid. (b) 263 122 Why ATP is energy currency of cell? Explain. 15. (a) (4 + 2) What are cardioglycosides? (b) How are lipids classified? Give an example for each class. 16. (a) (4 + 2)What is PUFA? Give an example. (b) Explain the Singer and Nicolson model of biological membrane. 17. (a) (4 + 2)Give the biological importance of cholic acid. (b) Mention the biological role of (i) cholesterol (ii) micelles. (a)
- 18.
 - (4 + 2)What is meant by rancidity? (b)
- What are Prostaglandins? Mention their biological importance. 19. (a)
 - (4 + 2)Write the structure of phosphatidyl ethanolamine. (b)
- How does an amino acid reacts with Ninhydrin and Formaldehyde? 20. (a)
 - (4 + 2)What is meant by peptide bond? (b)
- Write a note on forces involved in the stabilization of tertiary structure of 21. (a) proteins.
 - (4 + 2)What are essential amino acids? Give an example. (b)

61513

- 22. (a) Mention the types of secondary structure of proteins and describe any one of them.
 - (b) What are exergonic reactions? Give an example.

(4 + 2)

- 23. (a) Explain the salient features of oxidative phosphorylation.
 - (b) Mention any two aromatic amino acids.

(4 + 2)

- 24. (a) Write the role of the following components in ETC:
 - (i) FMN
 - (ii) NAD+
 - (iii) Ubiquinone
 - (iv) Cytochrome C.
 - (b) Give the structure of Coenzyme Q.

(4 + 2)

- 25. (a) What are High energy compounds? Mention three examples.
 - (b) Mention the number of ATP molecules produced by NADH and FADH₂.

(4 + 2)

